

# A biogenetically patterned enantiospecific synthesis of allopupukeanones

A. Srikrishna\* and G. Satyanarayana

Department of Organic Chemistry, Indian Institute of Science, Bangalore 560012, India

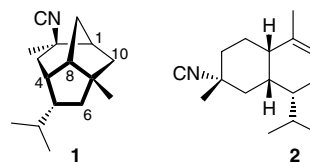
Received 28 August 2005; revised 23 October 2005; accepted 2 November 2005

Available online 17 November 2005

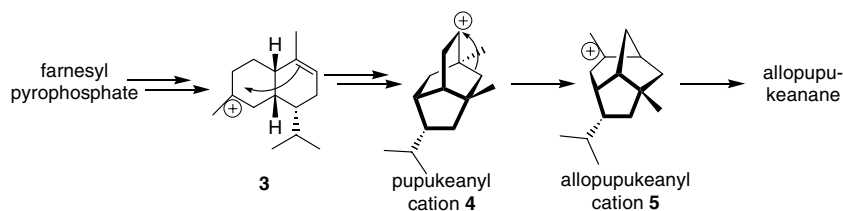
**Abstract**—The first enantiospecific synthesis of allopupukeanones has been accomplished starting from 6-methylcarvone. A biogenetically patterned rearrangement of a pupukeanane to allopupukeanane was employed as the key step.

© 2005 Elsevier Ltd. All rights reserved.

Nudibranchs are well known for various types of secondary metabolites in their mucous secretions as a part of self-defence. They have therefore been the focus of many marine natural product studies.<sup>1,2</sup> Members of the genus *Phyllidia* have gained the reputation of being especially pungent due to the presence of sesquiterpene isonitriles. In this context, the research group of Fusetani investigated *P. pustulosa* collected from Hachijo-Jima island. Bioassay-guided investigations led to the isolation of two new sesquiterpene isonitriles, 2-isocyanopallopupukeanane **1** and 4 $\alpha$ -isocyno-9-amorphene **2**, whose structures were deduced by extensive 2D NMR spectral analysis.<sup>3</sup> 2-Isocyanopallopupukeanane **1** has shown ichthyotoxicity towards the killifish *Oryzias latipes*. Biogenetically, the origin of pupukeananes<sup>4</sup> and allopupukeananes<sup>7</sup> can be explained by a common pathway via cyclisation and rearrangement of the bicyclic sesquiterpenes, cadinanes.<sup>2,3</sup> As depicted in Scheme 1, cyclisation and rearrangement of the carbocation **3**, derived from amorphene (a cadinane), generates the pupukeanyl cation **4**, which undergoes a 1,2-alkyl shift generating the allopupukean-2-yl carbocation **5**.



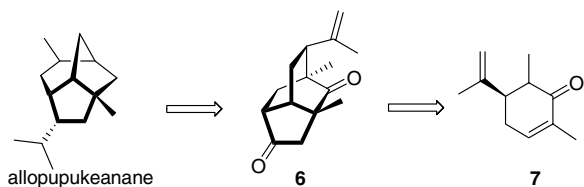
The presence of an interesting tricyclo[5.2.1.0<sup>4,8</sup>]decane carbon framework incorporating two quaternary carbon atoms, six stereogenic carbon atoms and an isonitrile functionality make 2-isocyanopallopupukeanane **1** an attractive and challenging synthetic target.<sup>4</sup> Herein, we describe the first enantiospecific synthesis of allopupukeanones, employing a biogenetically patterned conversion of a pupukeanane to an allopupukeanane as the key step. The retrosynthetic analysis is depicted in Scheme 2. By utilising an isopropenyl moiety as a masked hydroxy group and the biogenetically patterned conversion of an isotwistane into a tricyclo[5.2.1.0<sup>4,8</sup>]decane,<sup>5</sup> isotwistane dione **6** was conceived as the key intermediate, which could be obtained from 6-methylcarvone **7**.



Scheme 1.

**Keywords:** Allopupukeananes; Pupukeananes; Molecular rearrangement.

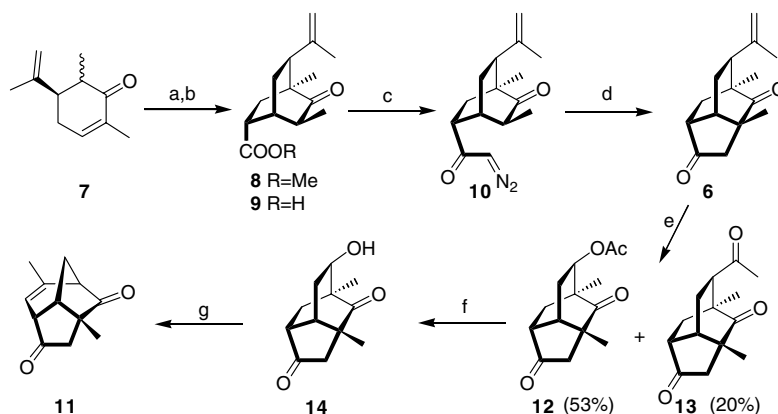
\* Corresponding author. Fax: +91 80 23600529; e-mail: [ask@orgchem.iisc.ernet.in](mailto:ask@orgchem.iisc.ernet.in)



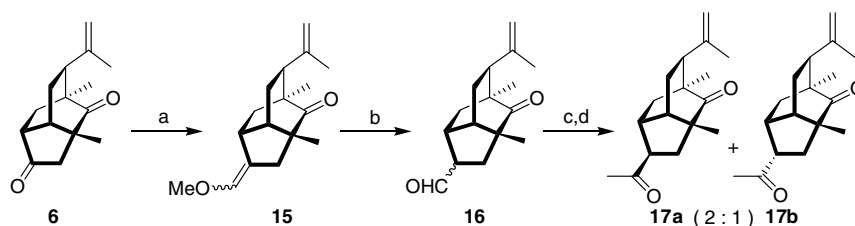
Scheme 2.

The synthetic sequence was initiated with the construction of the isotwistane dione **6** employing the methodology developed earlier in our laboratory (Scheme 3).<sup>6</sup> Thus, reaction of 6-methylcarvone **7** with lithium hexa-

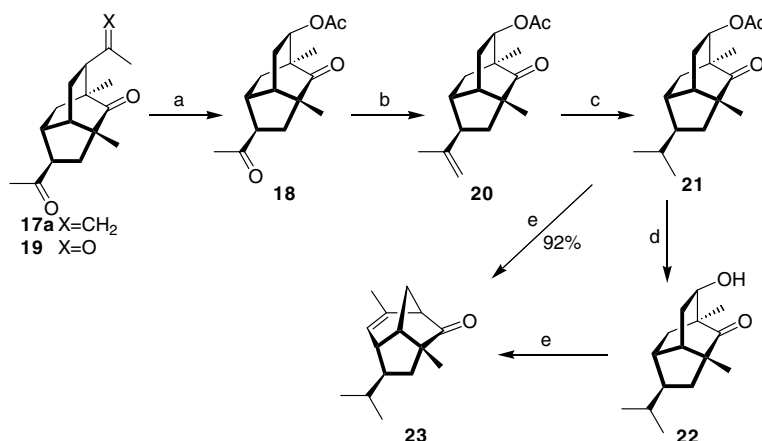
methylsilazide (LiHMDS) followed by treatment of the resultant kinetic dienolate with 1.1 equiv of methyl acrylate furnished bicyclo[2.2.2]octanecarboxylate **8** via intermolecular Michael addition followed by intramolecular Michael addition, which on base catalysed hydrolysis furnished the acid **9**. Reaction of the acid **9** with oxalyl chloride followed by treatment of the resultant acid chloride with diazomethane generated the diazoketone **10**. Reaction of the diazoketone **10** with a catalytic amount of rhodium trifluoroacetate in refluxing methylene chloride furnished isotwistane dione **6**, in 50% yield, via regioselective C–H insertion of the intermediate rhodium carbenoid.



**Scheme 3.** Reagents, conditions and yields: (a) LiHMDS, hexane,  $\text{CH}_2=\text{CHCOOMe}$ ,  $-10^\circ\text{C} \rightarrow \text{rt}$ , 3 h, 75%; (b) NaOH, MeOH, reflux, 8 h, 93%; (c) (i)  $(\text{COCl})_2$ ,  $\text{C}_6\text{H}_6$ , rt, 2 h; (ii)  $\text{CH}_2\text{N}_2$ ,  $\text{Et}_2\text{O}$ ,  $0^\circ\text{C} \rightarrow \text{rt}$ , 3 h; 90%; (d)  $\text{Rh}_2(\text{tfa})_4$ ,  $\text{CH}_2\text{Cl}_2$ , reflux, 4 h, 50%; (e)  $\text{O}_3/\text{O}_2$ ,  $\text{CH}_2\text{Cl}_2\text{-MeOH}$  (9:1),  $-70^\circ\text{C}$ ;  $\text{Ac}_2\text{O}$ ,  $\text{Et}_3\text{N}$ , DMAP,  $\text{C}_6\text{H}_6$ , reflux, 6 h, 73%; (f)  $\text{K}_2\text{CO}_3$ , MeOH, rt, 10 h, 85%; (g) PTSA,  $\text{C}_6\text{H}_6$ , reflux, 3 h, 74%.



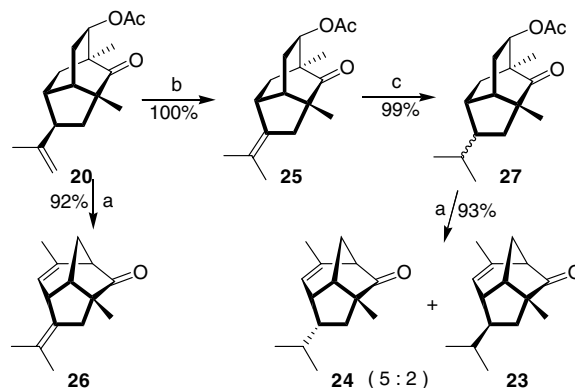
**Scheme 4.** Reagents, conditions and yields: (a)  $\text{Ph}_3\text{P}=\text{CHOMe}$ , THF, reflux, 3 h, 64%; (b) 3 N HCl, THF, rt, 1 h, 86%; (c)  $\text{MeMgI}$ ,  $\text{Et}_2\text{O}$ ,  $0^\circ\text{C} \rightarrow \text{rt}$ , 0.5 h, 96%; (d) PCC, silica gel,  $\text{CH}_2\text{Cl}_2$ , rt, 3 h, 95%.



**Scheme 5.** Reagents, conditions and yields: (a)  $\text{O}_3/\text{O}_2$ ,  $\text{CH}_2\text{Cl}_2\text{-MeOH}$  (9:1),  $-70^\circ\text{C}$ ;  $\text{Ac}_2\text{O}$ ,  $\text{Et}_3\text{N}$ ,  $\text{C}_6\text{H}_6$ , reflux, 5 h, 87% (**18:19** 3:1); (b)  $\text{Ph}_3\text{P}=\text{CH}_2$ ,  $\text{C}_6\text{H}_6$ , rt, 0.5 h, 75%; (c)  $\text{H}_2$ , 10% Pd-C, hexane, 1 atm, 6 h, 99%; (d)  $\text{K}_2\text{CO}_3$ , MeOH, rt, 10 h, 93%; (e) PTSA,  $\text{C}_6\text{H}_6$ , reflux, 3 h, 92%.

Conversion of isotwistane dione **6** into a tricyclo-[5.2.1.0<sup>4,8</sup>]decenone **11** was first investigated. Thus, ozonolysis of the isopropenyl group in the tricyclic dione **6** in methanol–methylene chloride at  $-70\text{ }^{\circ}\text{C}$  followed by treatment of the resultant methoxyhydroperoxide with acetic anhydride, triethylamine and a catalytic amount of DMAP in refluxing benzene furnished acetate **12**, along with the normal ozonolysis product, trione **13**. Hydrolysis of keto-acetate **12** with potassium carbonate in methanol gave alcohol **14** in 85% yield. Treatment of alcohol **14** with *p*-toluenesulfonic acid (PTSA) in refluxing benzene for three hours furnished cleanly the rearranged product **11** in 74% yield, whose structure was deduced from its spectral data.<sup>†</sup>

Subsequently, extension of the methodology for the synthesis of allopopukeananes was investigated (Schemes 4–6). For further elaboration of isotwistane dione **6**, the generation of an isopropyl group at the C-5 carbon was addressed prior to the degradation of the isopropenyl group. The relative steric crowding of the C-2 ketone when compared to that of the C-5 ketone group was exploited for the selective introduction of a side



**Scheme 6.** Reagents, conditions and yields: (a) PTSA, C<sub>6</sub>H<sub>6</sub>, reflux, 3 h; (b) RhCl<sub>3</sub>·*n*H<sub>2</sub>O, EtOH, reflux, 24 h; (c) H<sub>2</sub>, PtO<sub>2</sub>, EtOH, 1 atm, 48 h.

chain at the C-5 carbon of isotwistane dione **6**.<sup>8</sup> Accordingly, Wittig reaction of the tricyclic dione **6** with methoxymethylenetriphenylphosphorane in refluxing THF for 3 h furnished a 2:1 *E/Z* mixture of enol ether **15**, in 64% yield, which on hydrolysis with 3 N hydrochloric acid in 0.02 M THF solution furnished a 2:1 epimeric mixture of the aldehyde **16**. Reaction of aldehyde **16** with methylmagnesium iodide followed by oxidation of the resultant secondary alcohol with pyridinium chlorochromate (PCC) and silica gel in methylene chloride for three hours at room temperature furnished a 2:1 mixture of the epimeric diones **17a,b** in 91% yield, which were separated by column chromatography on silica gel. The structures of the two isomers **17a,b** were established from their spectral data.

First degradation of the isopropenyl group in the major dione **17a** was addressed via Creigee rearrangement.<sup>7</sup> Ozonolysis of the tricyclic dione **17a** in a mixture of methanol–methylene chloride followed by treatment of the intermediate methoxyhydroperoxide with a mixture of acetic anhydride, triethylamine and a catalytic amount of DMAP in refluxing benzene furnished the acetate **18**, along with a varying amount of the normal ozonolysis product, trione **19**. In order to establish the stereochemistry of the acetyl group unambiguously, diketo-acetate **18** was subjected to X-ray diffraction analysis (Fig. 1).<sup>9</sup> Wittig reaction of keto-acetate **18** with methylenetriphenylphosphorane furnished olefin **20**, which on hydrogenation in hexane at one atmospheric pressure of hydrogen (balloon) using 10% palladium on activated charcoal as the catalyst generated keto-acetate **21** in 75% yield, which on hydrolysis gave hydroxyketone **22** in 93% yield. Treatment of the hydroxyketone **22** with 1.5 equiv of PTSA in refluxing benzene for 3 h furnished the 5-epiallopopukean-10-one **23**, in 92% yield, in a highly regioselective manner.<sup>†</sup> Alternatively, it was found that the reaction of keto-acetate **21** with PTSA in refluxing benzene for 8 h also furnished 5-epiallopopukean-10-one **23**, directly.

Next, the synthesis of allopopukean-10-one **24**, starting from the tricyclic dione **17b**, was investigated. However, a comparable sequence led to a mixture of epimers at the C-5 carbon, due to epimerisation of the acetyl group

<sup>†</sup>Yields refer to isolated and chromatographically pure compounds. All the compounds exhibited spectral data (IR, <sup>1</sup>H and <sup>13</sup>C NMR and mass) consistent with their structures. Selected spectral data for (1*R*,4*R*,7*R*,8*S*)-2,7-dimethyltricyclo[5.2.1.0<sup>4,8</sup>]dec-2-ene-5, 10-dione **11**: Mp: 116–118 °C; [ $\alpha$ ]<sub>D</sub><sup>25</sup>:  $-165.5$  (*c* 2.0, CHCl<sub>3</sub>); IR (neat):  $\nu_{\text{max}}/\text{cm}^{-1}$  1738; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>+CCl<sub>4</sub>):  $\delta$  5.20 (1H, d, *J* 3.9 Hz), 3.19 (1H, br s), 2.84 (1H, dd, *J* 4.5 and 1.8 Hz), 2.61 (1H, t, *J* 5.1 Hz), 2.53 (1H, d, *J* 19.2 Hz, H-6A), 2.18 (1H, t of d, *J* 12.0 and 4.2 Hz, H-9A), 1.98 (1H, d, *J* 19.2 Hz, H-6B), 1.81 (3H, s, olefinic CH<sub>3</sub>), 1.66 (1H, d, *J* 12.0 Hz, H-9B), 1.27 (3H, s); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>+CCl<sub>4</sub>):  $\delta$  212.0 (C), 211.5 (C), 139.6 (C, C-2), 117.3 (CH, C-3), 54.3 (CH), 51.6 (C, C-7), 51.5 (CH), 44.0 (CH<sub>2</sub>), 43.9 (CH), 26.3 (CH<sub>2</sub>), 23.3 (CH<sub>3</sub>), 22.2 (CH<sub>3</sub>); Mass: *m/z* 190 (M<sup>+</sup>, 70%), 162 (13), 147 (25), 134 (27), 119 (100), 105 (45), 98 (11), 93 (38), 92 (75), 91 (77); HRMS: *m/z* Calcd for C<sub>12</sub>H<sub>14</sub>O<sub>2</sub>Na (M+Na): 213.0891. Found: 213.0901. Anal. Calcd for C<sub>12</sub>H<sub>14</sub>O<sub>2</sub>: C, 75.76; H, 7.42%. Found: C, 75.70; H, 7.42%. For (1*R*,4*R*,5*S*,7*R*,8*S*)-5-isopropyl-2,7-dimethyltricyclo[5.2.1.0<sup>4,8</sup>]dec-2-en-10-one (5-epiallopopukean-2-en-10-one **23**): [ $\alpha$ ]<sub>D</sub><sup>24</sup>:  $+321.8$  (*c* 4.8, CHCl<sub>3</sub>); IR (neat):  $\nu_{\text{max}}/\text{cm}^{-1}$  1737; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>+CCl<sub>4</sub>):  $\delta$  5.16 (1H, br s, H-3), 2.85 (1H, br s), 2.54 (1H, dd, *J* 4.5 and 1.5 Hz), 2.24 (1H, t, *J* 4.8 Hz), 2.00–1.82 (2H, m), 1.70 (3H, s, olefinic CH<sub>3</sub>), 1.62 (1H, d, *J* 11.7 Hz), 1.60–1.40 (2H, m), 1.36 (1H, dd, *J* 14.4 and 2.7 Hz), 1.12 (3H, s, *tert* CH<sub>3</sub>), 0.92 and 0.91 (6H, 2 × d, *J* 6.3 Hz, CH(CH<sub>3</sub>)<sub>2</sub>); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>+CCl<sub>4</sub>):  $\delta$  215.7 (C, C-10), 133.8 (C, C-2), 127.7 (CH, C-3), 54.5 (C), 51.3 (CH), 51.0 (CH), 46.8 (CH), 45.9 (CH), 39.0 (CH<sub>2</sub>), 31.6 (CH), 28.0 (CH<sub>2</sub>), 22.8 (2C, CH<sub>3</sub>), 21.6 (CH<sub>3</sub>), 21.1 (CH<sub>3</sub>); Mass: *m/z* 218 (M<sup>+</sup>, 15%), 190 (13), 147 (100), 119 (28), 105 (77); HRMS: *m/z* Calcd for C<sub>15</sub>H<sub>23</sub>O (M+H): 219.1749. Found: 219.1756. (1*R*,4*R*,5*R*,7*R*,8*S*)-5-Isopropyl-2,7-dimethyltricyclo[5.2.1.0<sup>4,8</sup>]dec-2-en-10-one (allopopukean-2-en-10-one **24**): [ $\alpha$ ]<sub>D</sub><sup>24</sup>:  $+325.0$  (*c* 1.0, CHCl<sub>3</sub>); IR (neat):  $\nu_{\text{max}}/\text{cm}^{-1}$  1736; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>+CCl<sub>4</sub>):  $\delta$  5.35 (1H, br s), 2.92 (1H, br s), 2.63 (1H, dd, *J* 4.5 and 1.8 Hz), 2.21 (1H, t, *J* 4.5 Hz), 1.89 (1H, t of d, *J* 11.7 and 4.5 Hz), 1.82–1.51 (4H, m), 1.70 (3H, s), 1.40–1.25 (1H, m), 1.14 (3H, s), 0.98 (3H, d, *J* 6.6 Hz), 0.83 (3H, d, *J* 6.6 Hz); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>+CCl<sub>4</sub>):  $\delta$  217.7 (C, C-10), 136.1 (C, C-2), 121.9 (CH, C-3), 53.3 (C, C-7), 52.3 (CH), 51.5 (CH), 49.4 (CH), 44.8 (CH), 41.7 (CH<sub>2</sub>), 31.2 (CH), 27.7 (CH<sub>2</sub>), 22.9 (CH<sub>3</sub>), 22.6 (CH<sub>3</sub>), 22.0 (CH<sub>3</sub>), 21.7 (CH<sub>3</sub>); Mass: *m/z* 218 (M<sup>+</sup>, 27%), 190 (13), 175 (6), 148 (23), 147 (100), 119 (43), 107 (52), 105 (66), 93 (16), 91 (21); HRMS: *m/z* Calcd for C<sub>15</sub>H<sub>23</sub>O (M+H): 219.1749. Found: 219.1755.

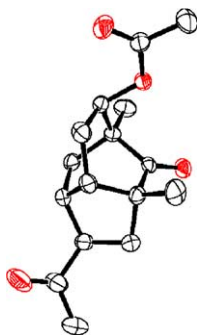


Figure 1. ORTEP diagram of **18**.

during the Wittig reaction. Hence, an alternative strategy via generation of the tetrasubstituted olefin **25** followed by hydrogenation was explored. Reaction of the keto-acetate **20** with PTSA in refluxing benzene for three hours led to isomerization of the double bond as well as the rearrangement of the ring system and generated allopupukeadienone **26** in 92% yield. On the other hand, reaction of keto-acetate **20** with rhodium chloride hydrate<sup>10</sup> in refluxing ethanol isomerised the olefin moiety and furnished the isopropylidene compound **25** in quantitative yield, which on hydrogenation with platinum oxide as the catalyst in ethanol furnished a 5:2 epimeric mixture of ketoacetate **27**. Treatment of the epimeric mixture of **27** with PTSA in refluxing benzene furnished a 5:2 mixture of allopupukeanones **24** and **23**, which were separated by column chromatography on 10% silver nitrate impregnated silica gel. The structures of allopupukean-10-ones **24** and **23** were established from their spectral data.

In summary, the first enantiospecific synthesis of allopupukeananes has been accomplished. A biogenetically patterned conversion of a pupukeanane to allopupukeanane was employed as the key step.

#### Acknowledgements

We thank the Council of Scientific and Industrial Research, New Delhi, for the award of a research fellowship to G.S.

#### References and notes

- Burreson, B. J.; Scheuer, P. J.; Finer, J. S.; Clardy, J. *J. Am. Chem. Soc.* **1975**, *97*, 4763–4764; Hagadone, M. R.; Burreson, B. J.; Scheuer, P. J.; Finer, J. S.; Clardy, J. *Helv. Chim. Acta* **1979**, *62*, 2484–2494.
- Karuso, P.; Poiner, A.; Scheuer, P. J. *J. Org. Chem.* **1989**, *54*, 2095–2097; Pham, A. T.; Ichiba, T.; Yoshida, W. Y.; Scheuer, P. J.; Uchida, T.; Tanaka, J.-i.; Higa, T. *Tetrahedron Lett.* **1991**, *32*, 4843–4846; He, H.-y.; Salva, J.; Catalos, R. F.; Faulkner, D. J. *J. Org. Chem.* **1992**, *57*, 3191–3194.
- Fusetani, N.; Wolstenholme, H. J.; Matsunaga, S.; Hirota, H. *Tetrahedron Lett.* **1991**, *32*, 7291–7294.
- To the best of our knowledge, there is only one approach reported in the literature on the synthesis of a racemic allopupukeanane, see: Ho, T.-L.; Kung, L.-R.; Chein, R.-J. *J. Org. Chem.* **2000**, *65*, 5774–5779.
- For acid catalysed conversion of isotwistanols to tricyclo[5.2.1.0<sup>4,8</sup>]decenes, Srikrishna, A.; Satyanarayana, G.; Kumar, P. R. *Tetrahedron Lett.* **2005**, *46*, in this issue. DOI:10.1016/j.tetlet.2005.11.008.
- Srikrishna, A.; Kumar, P. R.; Gharpure, S. J. *Tetrahedron Lett.* **2001**, *42*, 3929–3931.
- Schreiber, S. L.; Liew, W.-F. *Tetrahedron Lett.* **1983**, *24*, 2363–2366.
- Since addition of isopropyl Grignard reagent or the isopropylidene Wittig reaction were unsuccessful, a longer route was employed.
- Crystal data for **18**: X-ray data were collected at 293 K on a SMART CCD-BRUKER diffractometer with graphite monochromated Mo-K $\alpha$  radiation ( $\lambda = 0.7107$  Å). Structure was solved by direct methods (SIR92). Refinement was by full-matrix least-squares procedures on F<sup>2</sup> using SHELXL-97. The non-hydrogen atoms were refined anisotropically whereas hydrogen atoms were refined isotropically. C<sub>16</sub>H<sub>22</sub>O<sub>4</sub>, MW = 278.34, Crystal system: monoclinic, space group: P2(1); cell parameters:  $a = 8.219(2)$  Å,  $b = 11.331(3)$  Å,  $c = 8.314(2)$  Å,  $\alpha = 90^\circ$ ,  $\beta = 101.28(4)^\circ$ ,  $\gamma = 90^\circ$ ,  $\rho = 759.2(3)$  Å<sup>3</sup>;  $Z = 2$ ;  $D_c = 1.218$  g cm<sup>-3</sup>;  $F(000) = 300$ ,  $\mu = 0.086$  mm<sup>-1</sup>. Total number of l.s. parameters = 185;  $R1[I > 2\sigma(I)] = 0.0424$  for 2647.  $Rw[I > 2\sigma(I)] = 0.1028$ ; GOF = 1.048; restrained GOF = 1.048 for all data. ORTEP diagram is depicted in Fig. 1 (for clarity hydrogen atoms were omitted). Crystallographic data (without structure factors) have been deposited with the Cambridge Crystallographic Data Centre and the depository number is CCDC 275775.
- Grieco, P. A.; Nishizawa, M.; Marinovic, N.; Ehmann, W. *J. Am. Chem. Soc.* **1976**, *98*, 7102–7103.